California Sate University Channel Islands

NEW COURSE PROPOSAL

Courses must be submitted by November 3, 2008, for priority catalog review.

DATE (Change if modified and redate file with current date))

10.3.08 REV 10.20.08

Course Level Information:

PROGRAM AREA(S)

CHEMISTRY

Course Information. [Follow accepted catalog format.]

Prefix(es) CHEM and Course No. 111

Title: CHEMISTRY OF LIFE - PROBLEM SOLVING Units: 1

Repeatability:

Prerequisites

Grading Scheme:

x Corequisites CHEM 110 – concurrent enrollment required

Consent of Instructor Required for Enrollment

Catalog Description (Do not use any symbols): An instructor/peer-supervised interactive problem-solving session for students in CHEM 110 where students work in small groups on problems related to the content in CHEM 110.

x A-F Grades		Repeatable	for a maximum of	f x Under	rgraduate				
		units							
Credit/No Credit		Total Completions Allowed Post-Baccalaureate/Credential							
Optional (Student Choice)		Multiple Enrollment in Same Semester Graduate							
1 ,	,	1							
Mode of Instruction/Components (Hours per Unit are defaulted).									
		Hours	Benchmark	Graded	CS & HEGIS #				
		per	Enrollment	Component	(Filled in by the Dean)				
	Units	Ūnit		-					
Lecture		1							
Seminar	1	1	24	<u> </u>					
Laboratory		3							
Activity		2							
Field		·							
Studies									
Indep Study									
Other Blank				_					

Leave the following hours per week areas blank. The hours per week will be filled out for you.

1 hours seminar per week

hours blank per week

2. Course Attributes:

General Education Categories: All courses with GE category notations (including deletions) must be submitted to the GE website: http://summit.csuci.edu/geapproval. Upon completion, the GE Committee will forward your documents to the Curriculum Committee for further processing.

A (English Language, Communication, Critical Thinking)

A-1 Oral Communication

A-2 English Writing

A-3 Critical Thinking

B (Mathematics, Sciences & Technology)

B-1 Physical Sciences

B-2 Life Sciences - Biology

B-3 Mathematics – Mathematics and Applications

B-4 Computers and Information Technology

C (Fine Arts, Literature, Languages & Cultures)

C-1 Art

C-2 Literature Courses

C-3a Language

		C-3b Multicultural								
		D (Social Perspectives)								
		E (Human Psychological and Physiological Perspectives)								
		UDIGE/INTD Interdisciplinary								
		Meets University Writing Requirement								
		Meets University Language Requirement								
		American Institutions, Title V Section 40404: Government US Constitution US History								
R	Refer to website, Exec Order 405, for more information: http://senate.csuci.edu/comm/curriculum/resources.htm									
	Service Learning Course (Approval from the Center for Community Engagement must be received before you									
c	an r	request this course attribute).								

- 3. Justification and Requirements for the Course. (Make a brief statement to justify the need for the course)
- A. Justification: This course is an optional problem-solving session for the Chemistry of Life course (CHEM 110) and provides students with an interactive, problem-solving session where students work in small teams to solve problems related to the course. Its function is to increase student success in the chemistry of life course, so that students have a lower likelihood of needing to repeat this course. CHEM 110 a requirement for students in the B.S. Nursing.
 - B. Degree Requirement:

 Requirement for the Major/Minor

 Elective for the Major/Minor

 Note: Submit Program Modification if this course changes your program.
- **4.** Learning Objectives. (List in numerical order)

These are the same as for CHEM 110.

Upon completion of the course, the student will be able to:

- 1) Describe the scientific method and how it is used to approach chemical problems
- 2) Explain the differences between elements, chemical compounds, ions, and mixtures
- 3) Calculate the concentrations and solubilities of compounds in mass percent and molarity
- 4) Define acids and bases and pH of solutions
- 5) Calculate hydrogen-ion concentration and pH
- 6) Discuss how and why acid-base reactions occur
- 7) Explain how and why oxidation-reduction reactions occur
- 8) Determine the rate of a reaction and the energy change in a reaction
- 9) Explain the molecular structure of inorganic, organic, and biological compounds
- 10) Describe fundamental nuclear chemical processes and their medical applications
- 11) Explain enzyme catalysis and inhibition
- 12) Describe energy production in the metabolism of sugars, proteins, and lipids
- 13) Define chemical hazards of particular classes of chemicals
- 14) Explain how chemicals interact with the human body

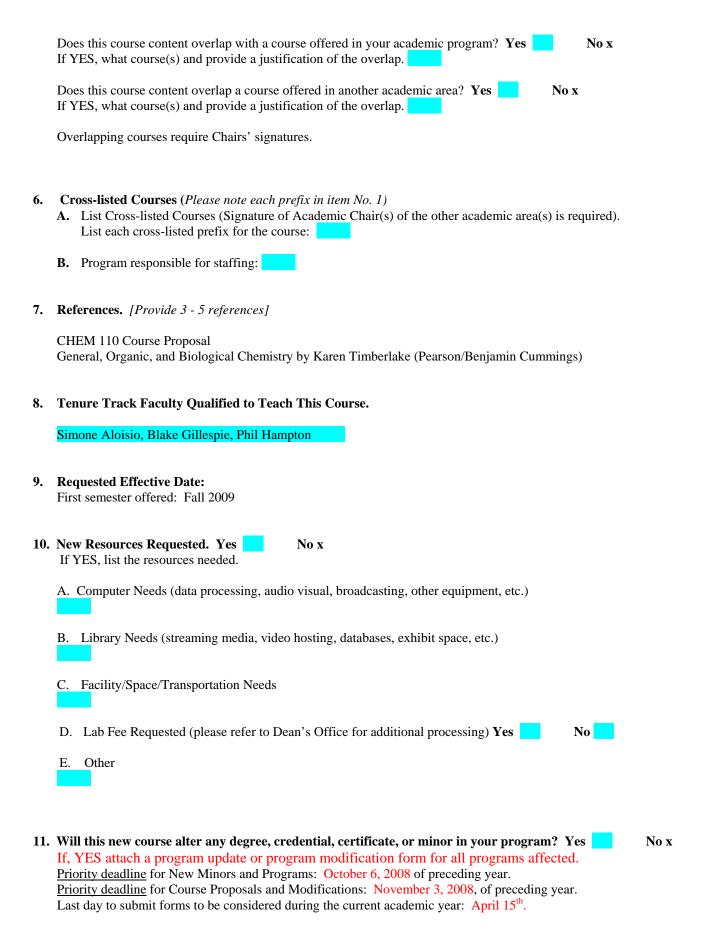
5. Course Content in Outline Form. [Be as brief as possible, but use as much space as necessary]

This is the same as CHEM 110

- I. Measurements and the Scientific Method
 - A. Units and Significant Figures
 - B. Unit Conversion
 - C. Scientific Method: Hypotheses, Theories, Experiments, and Conjecture
- II. Chemical Composition
 - A. Subatomic Particles, Atoms, and the Periodic Table
 - B. Molecules and the Nature of the Chemical Bonds
 - C. Compounds and Mixtures
 - D. Ions and Salts
 - E. Molecular Structure of Inorganic Compounds
- III. Physical Properties of Matter
 - A. States of Matter
 - B. Mass, Density, and Viscosity
 - C. Solubility and Solutions
 - D. Chemical Hazards of Gases, Liquids, and Solids
- IV. Chemical Reactions
 - A. Acid-Base Chemistry
 - B. Oxidation-Reduction Reactions
 - C. Rates of and Energy Changes in Reactions
 - D. Classifications of Chemical Reactions
 - E. Nuclear Chemistry and its Applications in Medicine
- V. Organic and Biological Molecules
 - A. Functional Groups and Interactions Between Molecules
 - B. Origin of Molecular Shape
 - C. Structures of Amino Acids, Sugars, Proteins, Nucleic Acids, and Lipids
 - D. Enzyme Catalysis and Inhibition
 - E. Amino Acid Function and Biosynthesis
 - F. Protein Function and Biosynthesis
 - G. Nucleic Acid Function and Biosynthesis
 - I. Energy Production: Metabolism of Sugars, Proteins, and Lipids
 - J. Biological Membranes: Structure, Function, Active and Passive Transport
 - K. Oxidative Phosphorylation and Electron-Transport

Note: Approximate coverage for this course is General Chemistry 40%, Organic Chemistry 20%, and Biochemistry 40%

 $9.15.08 \text{ km}^2$



Date

Proposer of Course (Type in name. Signatures will be collected after Curriculum approval)

Approval Sheet

Program/Course: CHEMISTRY/CHEM 111

If your course has a General Education Component or involves Center affiliation, the Center will also sign off during the approval process.

Multiple Chair fields are available for cross-listed courses.

Program Chair			
-	Signature	Date	
Program Chair			
	Signature	Date	
Program Chair			
	Signature	Date	
General Education Chair			
	Signature	Date	
Center for International Affairs Director			
	Signature	Date	
Center for Integrative Studies Director			
·	Signature	Date	
Center for Multicultural Engagement Director			
·	Signature	Date	
Center for Civic Engagement Director			
	Signature	Date	
Curriculum Chair			
	Signature	Date	
Dean of Faculty			
	Signature	Date	